Study of thermal expansion of mercury indium telluride crystals by XRD technique

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Abstract: The thermal expansion behavior of mercury indium telluride (MIT) crystals, Hg(3−3x)In2xTe3(x=0.5), based on X-ray diffraction experimental data is studied at 298−573 K. The variation of the lattice parameter of MIT crystals with temperature was determined and the thermal expansion coefficient was deduced to be 6.18×10−6 K−1. The results of the thermal expansion are fitted to polynomial expressions. It is found that the lattice parameter decreases quickly with temperature increasing at 298−330 K and then increases continuously up to 573 K. The minimum lattice parameter corresponds to a maximum shrinkage of 0.06%.

Key words: mercury indium telluride; crystal growth; XRD; thermal expansion; semiconductor material

1 Introduction

Solid solutions of II-VI/III-VI system attract many scientists due to their special feature of the high concentration of the structural vacancies[1−4]. Mercury indium telluride (MIT) with the formula of Hg(3−3x)In2xTe3 in HgTe/In2Te3 system is disclosed to be one of the important novel II-VI/III-VI compounds. For Hg(3−3x)In2xTe3, three stable phases are known at x=0.375, x=0.5 and x=0.75 called β, α2 and γ, respectively[5]. β phase has a body-centered tetragonal structure. It undergoes a transition from ordered structure to partially ordered structure at 720 K[6] and reacts peritectically with α2 and γ at 973 K[5]. α2 phase is face-centered cubic and its structure is related to that of the zinc-blende type ZnS. At 585 K, α2 undergoes an order-disorder transformation[7]. It has been reported that α2 completely decomposes to In2Te5 and In1Te3 with Hg escaped at 661 K at open condition[8]. The melting point of α2 is 982 K[9]. γ phase belongs to the chalcopyrite structure and melts congruently at 982 K[10]. Among them, MIT (x=0.5) is considered as a promising material for fast and efficient near-infrared photodetectors[11]. Information on thermal expansion coefficient and thermal conductivity of MIT as a function of temperature is essential to evaluate the thermophysical properties under temperature gradients. So far, no details concerning the thermal expansion behavior of MIT crystals studied by XRD technique were reported. Therefore, thermal expansion investigation was carried out for MIT (x=0.5) compound of HgTe-In2Te3 system. The aim of the present work is to study the thermal expansion behavior of MIT (x=0.5) compound as a function of temperature.

2 Experimental

The sample of MIT single crystal was grown by using the vertical Bridgman (VB) method[12−13]. The raw materials for crystal growth were synthesized by direct reaction in vacuum in a quartz ampoule using the high purity (99.999 99%) elements of Te, In and Hg. The crystal growth was performed in a vertical Bridgman furnace in the same ampoule. A temperature gradient of 10 K/cm and a growth rate of less than 1 mm/h were adopted. The ingot was in-situ annealed for 72 h after growth.

Phase identification and crystal structure of the as-grown crystals was examined by X-ray powder
diffraction (XRPD) method. In order to monitor the crystal structure of MIT material, high temperature X-ray diffraction (HT-XRD) measurements were carried out at 298–573 K. The HT-XRD patterns were taken at an interval of 50 K using a high temperature attachment. The sample cut from the as-grown crystal was crushed and finely ground into powder, then mixed with drops of acetone and well distributed on a Pt-Rh sample carrier, which is spot welded at the bottom with a Pt/Pt-10%Rh thermocouple to collect temperature signals. The programmed heating rate was set to 5 °C/min. An XPert MPO Pro X-ray diffractometer was employed to record the diffraction data in the 2θ range of 10°–120° with the step of 0.033°. The X-ray source was Cu Kα with the wavelength of 0.154 060 nm. The accelerating voltage was 40 kV and the current was 35 mA. Divergence slit and the receiving slit were 0.435 4° and 0.05°, respectively. The lattice parameters at each temperature were determined by using a computer DICVOL program with an accuracy of ±1.0×10⁻⁵ nm.

3 Results and discussion

The thermal expansion of a crystal implied by the thermal-expansion coefficient has great influence on crystal growth and on its possible applications. It is known that the thermal-expansion coefficient is a symmetrical second-rank tensor and can be geometrically described by a second-rank indicative surface, which can be expressed in the principal axis coordinate by the following form[14–15]:

\[
\alpha_{11}x_1^2 + \alpha_{22}x_2^2 + \alpha_{33}x_3^2 = 1
\]

where the semi-axis lengths of the three principal axes of the second-rank indicative surface are (\(\alpha_{11}\))⁻¹/², (\(\alpha_{22}\))⁻¹/², and (\(\alpha_{33}\))⁻¹/². The shape of the second-rank indicative surface is determined by the values of the principal components (\(\alpha_{11}\), \(\alpha_{22}\) and \(\alpha_{33}\)). The thermal expansion coefficient tensor with respect to the axes in the principal coordinate is

\[
\alpha = \begin{bmatrix}
\alpha_{11} & 0 & 0 \\
0 & \alpha_{22} & 0 \\
0 & 0 & \alpha_{33}
\end{bmatrix}
\]

For MIT (x=0.5) crystal, it has a defect zinc-blende structure and belongs to \(\text{4}3\text{m}\) point group[12], which has four triad axes along the diagonal of the crystallographic lattice and three diad axes along the crystallographic axes that are selected as the axes in the principal coordinate. According to the Neumann’s principle, the components of the thermal-expansion coefficient tensor can be calculated to be only one independent principal component (\(\alpha_{11}=\alpha_{22}=\alpha_{33}\)). The second-rank indicative surface of MIT crystal has three equal semi-axis lengths and is a sphere.

In order to determine the value of \(\alpha_{11}\), the thermal expansion of MIT crystal by HT-XRD method was measured. The HT-XRD patterns up to 573 K clearly display the shift in the X-ray diffraction line positions showing the change of the unit cell. The XRD pattern at room temperature of MIT crystal is shown in Fig.1, from which the unit cell parameters are determined to be 0.629 34 nm with a volume of 0.249 26 nm³. Fig.2 shows the variation of lattice parameters of MIT crystal with temperature at 298–573 K. It can be seen that the lattice
parameter of MIT crystal decreases with temperature at the first stage, then increases continuously up to 573 K. The volume of the unit cell has the same trend. The lattice parameter inclines quickly at a ratio of about $1.52 \times 10^{-6}$ nm/K in the shrinkage stage, and then begins to expand. At the expansion stage, the expansion ratio reduces to about $5.8 \times 10^{-6}$ nm/K. The thermal expansion behavior of MIT crystal exhibits a minimum in the expansion curve. The shrinkage is probably caused by removal of the residual thermal defects formed during the cooling process when MIT crystals are quenched.

The lattice parameter of MIT crystal against temperature can be probably expressed by a fourth-order polynomial relation obtained from the least squares method as implied by Eqns.(3)–(4):

$$a_T=0.70966-0.000747+0.25017 \times 10^{-3} T^2$$

$$+0.36585 \times 10^{-6} T^3+0.19774 \times 10^{-9} T^4$$  \hspace{1cm} (3)

$$V_T=345.9411 \times 10^{-6} T^2-0.8942 \times 10^{-9} T^3$$

$$+2.3832 \times 10^{-12} T^4$$  \hspace{1cm} (4)

The theoretical temperature corresponding to the minimum lattice parameter was determined to be 330 K from the differential results of $\partial a/\partial T=0$. The values of the lattice parameters observed and calculated from Eqn.(3) are both given in Table 1. The total error of the individually observed lattice parameter is probably no more than 0.000 2 nm. Eqn.(3) is taken as a better representation of the results.

### Table 1 Lattice parameters of MIT crystals at different temperatures

<table>
<thead>
<tr>
<th>$T$/K</th>
<th>$a_{obs}$/nm</th>
<th>$a_{calc}$/nm</th>
<th>Difference/nm</th>
</tr>
</thead>
<tbody>
<tr>
<td>298</td>
<td>0.629 34</td>
<td>0.629 18</td>
<td>-0.000 16</td>
</tr>
<tr>
<td>323</td>
<td>0.628 96</td>
<td>0.628 90</td>
<td>-0.000 06</td>
</tr>
<tr>
<td>373</td>
<td>0.629 24</td>
<td>0.629 99</td>
<td>-0.000 25</td>
</tr>
<tr>
<td>423</td>
<td>0.629 52</td>
<td>0.629 40</td>
<td>-0.000 12</td>
</tr>
<tr>
<td>473</td>
<td>0.629 94</td>
<td>0.629 74</td>
<td>-0.000 20</td>
</tr>
<tr>
<td>523</td>
<td>0.630 16</td>
<td>0.629 93</td>
<td>-0.000 23</td>
</tr>
<tr>
<td>573</td>
<td>0.630 41</td>
<td>0.630 18</td>
<td>-0.000 23</td>
</tr>
</tbody>
</table>

The values of the thermal expansion coefficient can generally be obtained by the expression $\alpha = (1/a) (\partial a/\partial T)$ from the relation of lattice parameter $a$ and temperature $T$. Supposing that the lattice parameter of MIT crystal at 298 K is $a_{298}$, the value of the thermal expansion coefficient ($\alpha_T$) can be calculated through the following equation:

$$\alpha_T = \frac{\Delta a}{\Delta T} \times \frac{1}{a_{298}}$$  \hspace{1cm} (5)

where $\Delta a$ is the variation of lattice parameter corresponding to a temperature change $\Delta T$. The value of the average thermal expansion coefficient at $298$–$573$ K can be theoretically calculated from Eqn.(5) to be $\alpha_T = 6.18 \times 10^{-6}$ K$^{-1}$. The volume coefficient of thermal expansion is equal to $18.54 \times 10^{-6}$ K$^{-1}$.

Fig.3 shows the thermal expansion coefficients of MIT crystal at 298–573 K. It is noticeable for the thermal expansion behavior of MIT crystal that there are two distinctive parts: shrinkage and expansion stages. The thermal expansion coefficient of the former is up to $-24.15 \times 10^{-6}$ K$^{-1}$, whereas that of the latter is $9.22 \times 10^{-6}$ K$^{-1}$.

![Fig.3 Thermal expansion coefficient of MIT (x=0.5) crystal](image)

The distinctive thermal expansion behavior suggests that in the process of MIT crystal growth, different cooling rates should be adopted; otherwise, a large amounts of defects are likely to be introduced into the crystal, such as microcracks.

### 4 Conclusions

1) The thermal expansion behavior of mercury indium telluride (MIT) crystals, Hg$_{x}$In$_{1-x}$Te$_3$ ($x=0.5$), grown by vertical Bridgman (VB) method was studied at 298–573 K by the high temperature X-ray diffraction (HT-XRD) measurements. It is found that the unit cell of MIT shrinks quickly with temperature increasing at 298–330 K and then expands continuously up to 573 K. There exists a maximum shrinkage of 0.06% at 330 K.

2) The lattice parameters of MIT crystal are fitted to a fourth-order polynomial expression as a function of temperature. There are two distinctive parts in the expansion curve of MIT crystals: shrinkage and expansion stages. The thermal expansion coefficient at the former stage is up to $-24.15 \times 10^{-6}$ K$^{-1}$, whereas that at the latter stage is $9.22 \times 10^{-6}$ K$^{-1}$. The average thermal expansion coefficient is $6.18 \times 10^{-6}$ K$^{-1}$ at 298–573 K.

### References


[5] SPENCER P M, RAY B. Phase diagram of the alloy system \( \text{Hg}_2\text{Te}_3-\text{In}_2\text{Te}_3 \) [J]. Brit J Appl Phys (J Phys D), 1968, 1: 299–301.


[10] WEITZE D, LEUTE V. The phase diagrams of the quasibinary systems \( \text{HgTe}\text{In}_2\text{Te}_3 \) and \( \text{CdTe}\text{In}_2\text{Te}_3 \) [J]. Journal of Alloys and Compounds, 1996, 236: 229–235.


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