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Crystalline structure and morphology of X phase in Cu-Al-Ni-Mn-Ti alloy^①

WANG Ming-pu(汪明朴), YANG Jie(杨杰), LI Zhou(李周),
WANG Yan-hui(王艳辉), WANG Zhi-wei(王志伟)

(Department of Materials Science and Engineering, Central South University,
Changsha 410083, P.R.China)

[Abstract] The composition, morphology, crystalline structure and formation and evolution of X phase in a Cu-12.3Al-2Ni-2Mn-1Ti alloy were studied. The results show that the X phase is a Ti-rich phase with atomic ratio of (Cu+Ni):Ti:Al = 2:1:1 and DO₃ or L₂ structure; it is directly formed in the liquid phase by crystallization in the process of solidification of the alloy; there forms free particle X phase through the progressive dissolution and breakdown of the inter-dendritic microstructure in the following homogenization process. The X phase has three different morphologies, i.e. X_L, X_{LS} and X_S, whose contents in the matrix rely on the heat treatment conditions.

[Key words] Cu-Al-Ni-Mn-Ti alloy; X phase; crystalline structure; morphology

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1 INTRODUCTION

Copper base shape memory alloys have low costs and good shape memory properties. Their shape memory effects originate from the thermoelastic martensitic transformations. A lot of work has been done on the martensitic transformations^[1,2], martensitic structures^[3~5], ageing effects^[6~8] and wide hysteresis effects^[9] and so on. In the recent years, the additions of minor alloying elements into the copper base shape memory alloys, especially the additions of Mn and Ti into the Cu-Al-Ni alloys^[10~14], have been drawing great attentions. The additions of elements Mn and Ti will help improve the properties of these alloys, especially there will form a dispersive secondary phase called X phase by adding Ti element. Much work has been done on the X phase in the Cu-Al-Ni-Ti alloys. Adachi et al^[10] believed that the X_L is precipitated from the β phase, X_S is precipitated from the supersaturated matrix on quenching, and X phase is a soft phase. Chang et al^[13] believed that the X phase is formed by the prior crystallization in the liquid phase at the initial stage of solidification, the X_L is formed by the spheroidizing of the X phase during the solid solution treatment in the high temperature β phase zone, and the X_S is precipitated by the nucleation and growth way. But Zhu et al^[14] believed that the X_L is produced through the following peritectic reaction: $L + \beta \rightarrow X_L$. By taking Cu-12.3Al-2Ni-2Mn-1Ti alloy as an example, the composition, morphology, crystalline structure, and formation and evolution of the X phase were studied in this paper.

2 ALLOY PREPARATION AND EXPERIMENTAL METHODS

The nominal composition of the tested alloy is Cu-12.3Al-2Ni-2Mn-1Ti in mass fraction. After medium frequency induction melting in a graphite crucible, the alloy was cast into a slab ingot using a permanent mould. After homogenization treatment at 900 °C for 4 h, the slab ingot was hot rolled into 1 mm thick sheets. Specimens cut along the rolling direction were heat treated at 760 °C for 5 min, 960 °C for 2 h, 960 °C for 2 h then furnace cooled to 760 °C and held for 1 h, and finally cooled on an iron sheet after being taken out from the furnace. Microstructure observations and microhardness measurements were made and composition analysis were realized by X-ray electron probe microanalysis in an X-650 SEM. The TEM observations and phase analyses were made with an H800 TEM.

3 RESULTS AND DISCUSSION

3.1 Morphology, composition and crystalline structure

Fig.1 shows the typical bright field images of the tested Cu-Al-Ni-Mn-Ti alloy. It can be seen that the alloy matrix is martensite and the X phase is imbedded in it. The X phase has three different morphologies under the same treatment conditions. The X_L presents spheric shape with a size of several microns, the X_{LS} presents plum blossom like shape with a size of 300~600 nm (Fig.1(a)), and the X_S presents cubic or spheric shape with a size of 20~30 nm

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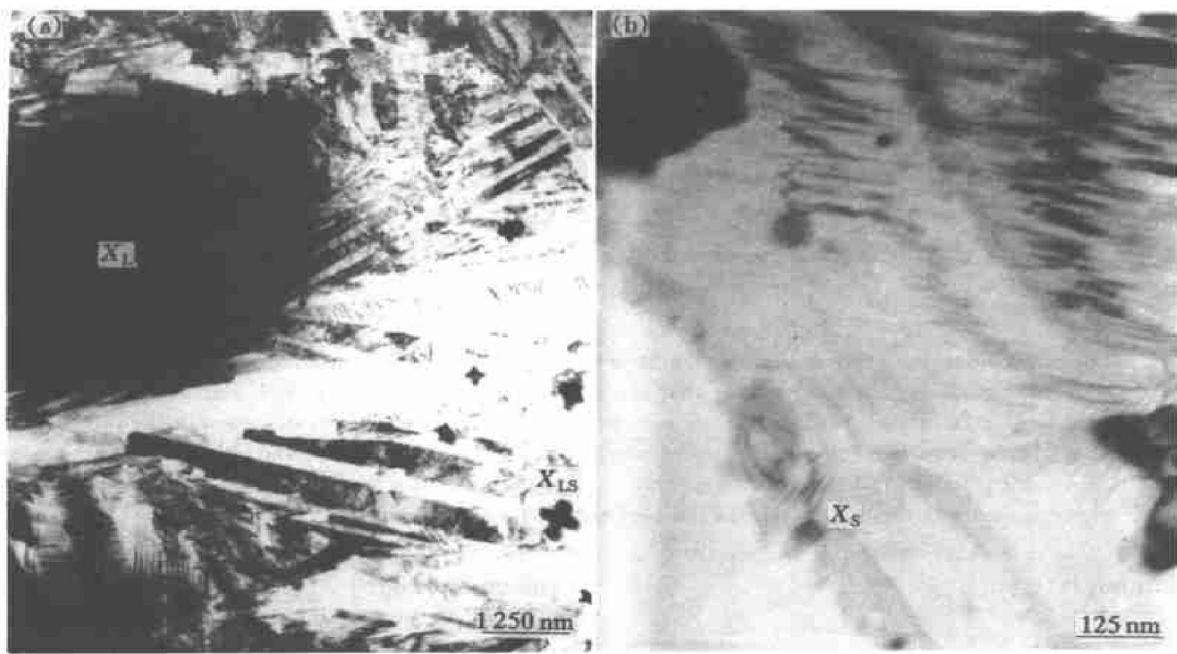


Fig.1 TEM images of Cu-Al-Ni-Ti alloy

(Fig.1(b)).

It is found by electron probe microanalysis of the specimens that the matrix martensite is poor in Ti, while the X phase is a Ti-rich phase with the composition of 35.5Cu-26.5Al-23Ti-15Ni in mole fraction which is close to (Cu+Ni): Ti:Al=2:1:1, i.e. the X phase satisfies the chemical formula of (Cu, Ni)₂TiAl.

Fig.2 shows the electron diffraction patterns of different X_L particles which can be determined to be the (100)^{*} and (331)^{*} crystal bands of the centered cubic lattice system. The lattice constant is calculated to be 6.06 Å or 5.89 Å respectively by using the two patterns. These values are different from the value of 5.978 Å reported in Ref.[10], which may be caused by the difference of the compositions of X phase in

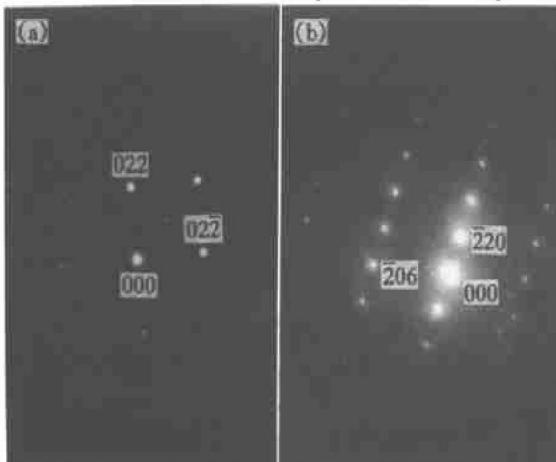


Fig.2 Electron diffraction patterns of X_L
(a) —(100)^{*}; (b) —(331)^{*}

this work and Ref.[10].

It can be deduced from the electronic diffraction patterns that the X_L possesses DO₃ or L₂₁ crystalline structure. The crystalline structure factor of the X_L can be expressed by

$$F_{HKL} = \{ f_a + f_b \exp[\pi i(H+K+L)/2] + f_a \exp[\pi i(H+K+L)] + f_c \exp[3\pi i(H+K+L)/2] \} \cdot \{ 1 + \exp[\pi i(H+K)] + \exp[\pi i(H+L)] + \exp[\pi i(K+L)] \} \quad (1)$$

It can be obtained from expression (1) that:

$$F = \begin{cases} 4(2f_a + f_b + f_c) H + K + L = 4n \\ 4(2f_a - f_b - f_c) H + K + L = 4n + 2 \\ 4(f_a - f_c) H + K + L = \text{odds} \end{cases} \quad (2)$$

These extinction effects agree well with the results of electronic diffractions, thus the previous judgment of the crystalline structure of the X phase is correct.

3.2 Formation and evolution of X phase

Fig.3 shows the evolution of microstructure in the homogenization process of the as-cast specimens. There is a large amount of network dendrites in as-cast microstructure and a little precipitation of α phase. It can be seen under high magnification microscope that the inter-dendritic microstructure is made of irregular deep-colored spheric particles connecting with each other. In the further homogenization process, the dendrites dissolve progressively and separate each other, and form many sphere-like particles dispersively distributed in the matrix. This situation will become stronger with increasing homogenization time. By means of electron probe microanalysis, it is

known that the inter-dendritic microstructure is a Ti-rich phase and with progressing homogenization, the compositions of those free particles separated from the parent phase will more approach to that of the X phase. Therefore it can be concluded that the dendritic microstructure crystallized at the initial stage of solidification is the prototype of the X phase.

Fig. 4 shows the evolution of microstructure of the hot-rolled sheets in the process of heat treatment. The deep-colored particles are the X phase; they are dispersively distributed in the matrix. After heat treatment at 760 °C for 5 min, the distribution of the X phase is more homogeneous, and the content of the X_L is larger than that of the X_{LS} . But after heat treatment at 960 °C for 2 h, the previous irregular X_L spheroidizes progressively, and reduces in content, while the content of the X_{LS} increases. After 960 °C, 2 h + 760 °C, 1 h treatment, the X_L becomes irregular in shape again and increases in content, but the content of the X_{LS} reduces. The above results indicate that the inter-dendritic microstructure in the as-cast alloy is the real resource of the X phase. That is to say, the X phase is formed in the liquid phase by crystallization in the process of solidification. When the temperature is lowered, the melt begins to solidify. Because the solid solubility of Ti in the matrix is small, it is easy to form compounds with the other elements and is considered "impurity" and discharged to the solid-liquid interface front, finally the Ti-rich

phase forms the inter-dendritic microstructure. In the following homogenization process, with the progress of the dissolution of the dendrites, the inter-dendritic microstructure is separated to form free X phase particles, which are dispersively distributed in the matrix and become more homogeneous after hot rolling.

There are several kinds of morphology of the X phase, and the main morphologies are X_L and X_{LS} . The finer X_S is difficult to observe by common optic microscopy. The contents of the X_L and X_{LS} depend heavily on the heat treatment conditions. The X phase is formed by the spheroidizing of the original Ti-rich X phase during the solid solution treatment in the high temperature phase zone. At the same time, the X_{LS} also begins to precipitate by the nucleation and growth way. If the cooling rate is large, then there may be no enough time for the X_{LS} to grow up, thus there occurs a large amount of X_{LS} . If the cooling rate is slow, there is some time for the X_{LS} to grow up, then there forms the X_L . But when directly treated at lower temperatures there only takes place the growth of the X_{LS} and forms the X_L , thus there also occurs the phenomenon of much X_L and little X_{LS} .

3.3 Hardness measurement of X_L

The hardnesses of the X_L and the matrix were measured to be Hv 351 and Hv 292, respectively. The X_L is harder compared with the matrix, which is

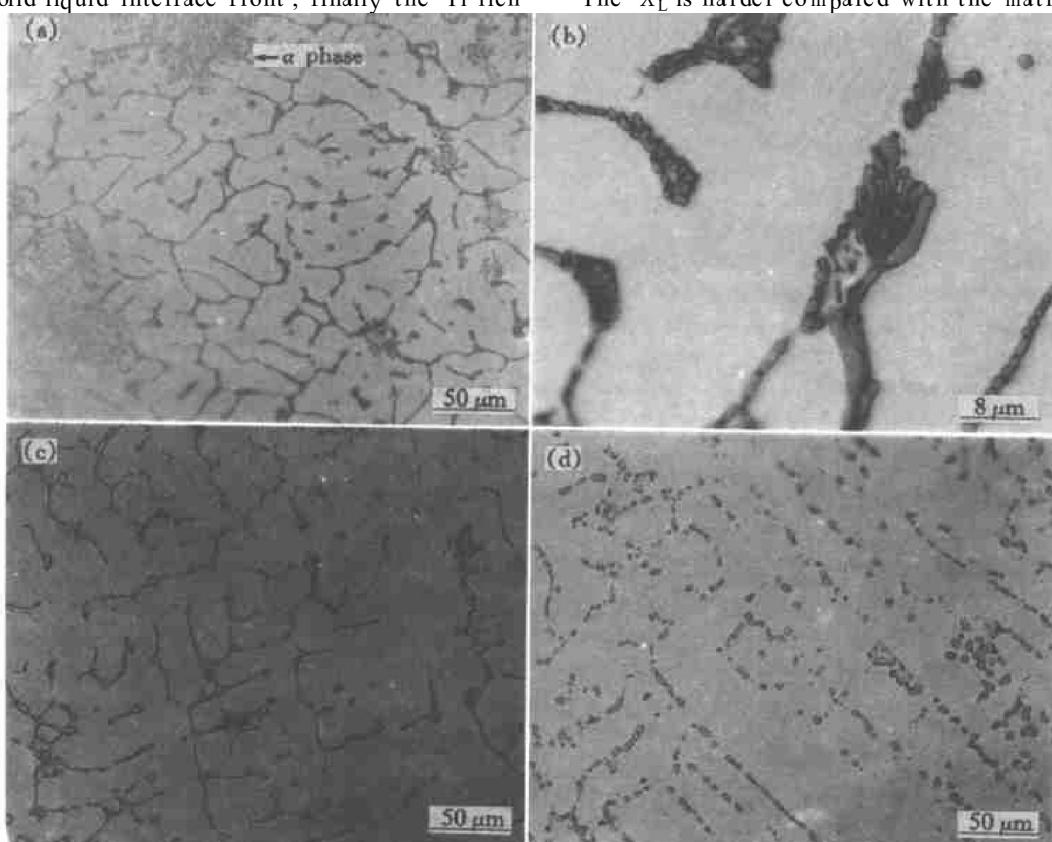


Fig.3 Evolution of microstructure of ingot during homogenization treatment
(a) and (b) — As-cast; (c) — 900 °C, 0.5 h; (d) — 900 °C, 4 h

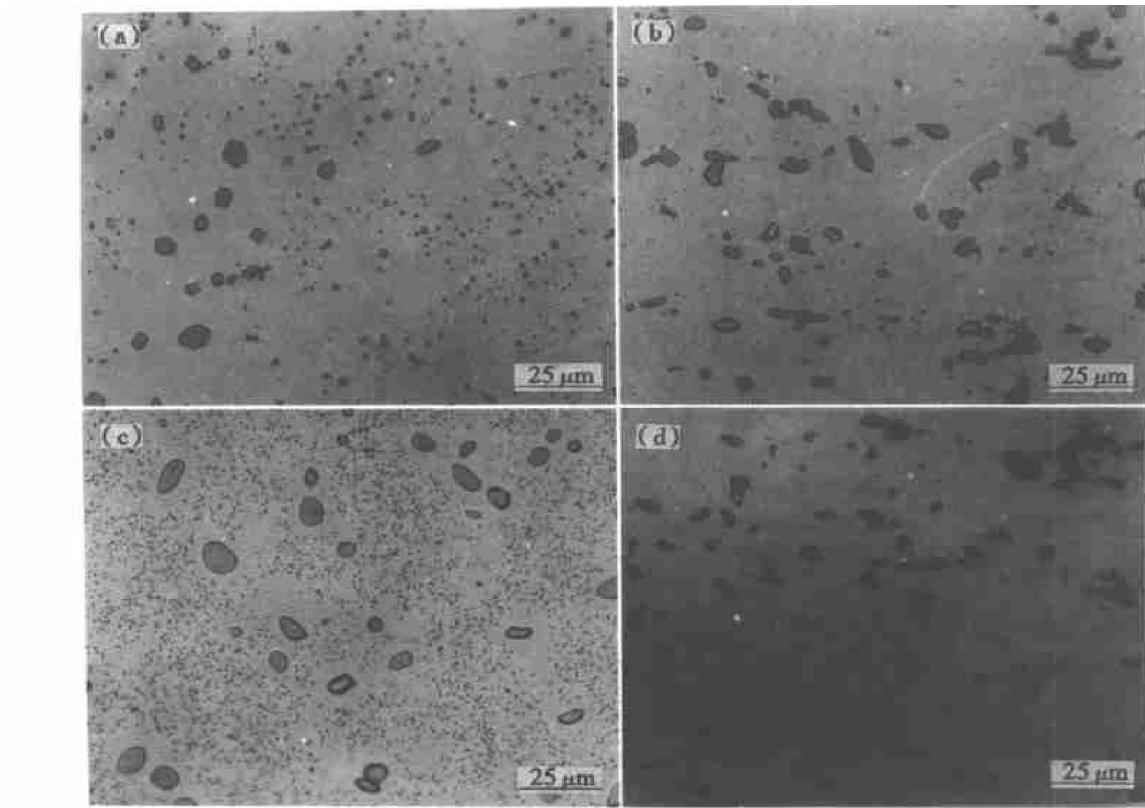


Fig.4 Evolution of microstructure of hot-rolled specimens during heat treatment
 (a) —As rolled ; (b) -760°C , 5 min ; (c) -960°C , 2 h ; (d) -960°C , 2 h \rightarrow 760°C , 1 h

different from the result reported in Ref.[10].

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