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Effect of annealing treatment on optical and electrical properties of ZnO films^①

WANG Wan-lu(王万录), LIAO Ke-jun(廖克俊), LI Li(李丽),
WU Zhi-hua(吴子华), WANG Yong-tian(王永田), ZHANG Jin(张津)
(Department of Applied Physics, Chongqing University, Chongqing 400044, China)

Abstract: The ZnO-Al films were prepared by R. F. magnetron sputtering system using a Zn-Al target (with purity of 99.99%). The obtained films were characterized by X-ray diffraction, SEM and optical and electrical measurements. The experimental results show that the properties of ZnO films can be further improved by annealing treatment. The crystallinity of ZnO films becomes better, and the optical gap energy is decreased, but thermoelectric power is enhanced after heat treatment. The optical gap energy decreases from 3.75 eV to 3.68 eV when the annealing temperature increases from 25 °C to 400 °C. This can be ascribed to the decrease of carrier concentration, resulting in Burstein shift.

Key words: ZnO films; annealing treatment; Seebeck effect; optical gap energy

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1 INTRODUCTION

ZnO is a II-VI compound semiconductor with a wide direct band gap and a hexagonal wurtzite structure (space group p6₃mc with cell parameters $a = 0.3249$ nm, $c = 0.5206$ nm). Owing to their unique physical and chemical properties, ZnO thin films can be used as surface acoustic wave devices and bulk acoustic resonators, transparent electrodes, gas and optical sensors, ultrasonic oscillators, transducers, optical wave guides, photoprotective coatings, ultraviolet laser, light-emitting diodes and light diodes^[1-7]. ZnO films have widely been studied by many research groups from synthesis techniques to actual applications. At present, ZnO films can be synthesized by various methods such as DC and RF magnetron sputtering^[8], PLD^[9], CVD^[10], SS-CVD^[11], spray pyrolysis^[12], sol-gel^[13]. P- and N-type doping ZnO films have also gained great progress^[14, 15]. However, some fundamental issues need to be further studied in order to obtain high quality ZnO thin films. In this work, effect of the annealing treatment on the optical and electrical properties of ZnO films was investigated. The experimental results show that the annealing treatment has an important influence on the structures and optical gap energy, and thermoelectric power of ZnO thin films. The results are very important for the quality and applications of ZnO films.

2 EXPERIMENTAL

ZnO thin films were prepared by R. F. Mag-

netron sputtering system using a Zn target (99.99%) containing Al of 1.5% in Zn with diameter of 7.62 cm and thickness of 5 mm. Soda lime glass was used as the substrates. The substrate materials were thoroughly cleaned with organic solvents and dried before loading in the deposition chamber. The reactive chamber was first pumped down to about 1.3332×10^{-6} Pa using molecular pump, and a sputter-etch of 5 min was removed the target surface contamination. After that, a mixture of argon and oxygen was introduced, and the ratio of argon to oxygen was controlled by two electronic flow controllers. The substrate temperature was monitored using a thermo-cooper attached near the substrate. The deposition conditions were shown in Table 1.

Table 1 Deposition conditions of ZnO films

Parameter	Value
R. F. Power/ W	150
R. F. frequency/ MHz	13.65
Substrate temperature/ °C	200
Pressure/ Pa	3×10^{-3}
Partial pressure of O ₂ / Pa	0.35
Deposition time/ min	45
Target-substrate distance/ mm	50

The ZnO films obtained were characterized by scanning electron microscopy, X-ray diffraction

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Correspondence: WANG Wan-lu, Professor; Tel: + 86-23-65103490; E-mail: wanluw@cqu.edu.cn

and optical and electrical measurements. SEM image of ZnO films is shown in Fig. 1. The ZnO particle size is about 100–200 nm, which depends on the deposition conditions.

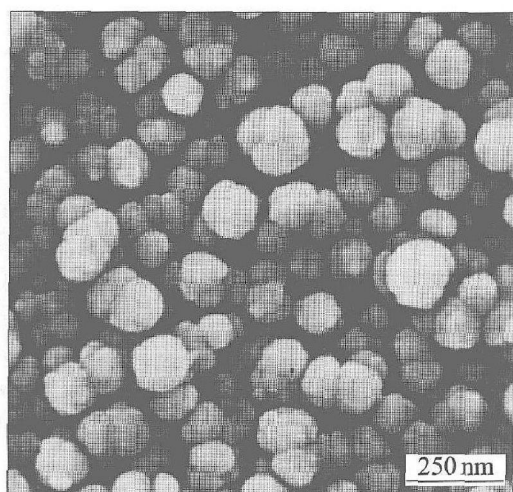


Fig. 1 SEM image of ZnO films

3 RESULTS AND DISCUSSION

ZnO films with 600 nm in thickness containing 1.5% Al concentration at room temperature were annealed in Ar atmosphere at temperature ranging from 25 °C to 400 °C. The heated rate of samples was about 10 °C/min, and time duration was 2 h. The experimental results indicate that the annealing treatment causes the X-ray diffraction peak to shift toward a high value as shown in Fig. 2. The peaks move from $2\theta = 34.7^\circ$ to $2\theta = 35.2^\circ$ when the annealing temperature increases from 25 °C to 400 °C as shown in Fig. 3. This is because of Al doping into ZnO films, resulting in a reduced lattice constant and internal stress degradation after heat treatment. On other hand, the crystallite size of ZnO also increases with annealing temperature in-

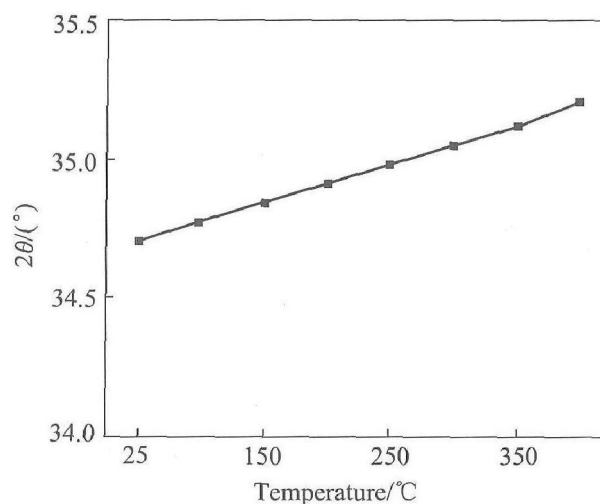


Fig. 3 Changes of XRD peak with annealing temperature

creasing. The particle size increases from 100 nm to 125 nm when the annealing temperature increases from 25 °C to 400 °C. From Fig. 2, it can be seen that the ZnO thin films show preferred orientation of (002) plane with hexagonal structure (wurtzite type).

The optical properties of ZnO films can be obtained by analyzing UV-Visible spectra. The film transmittance is given by the following formula^[16]:

$$T = (1 - R_1)(1 - R_2)(1 - R_3) \exp(-\alpha) \cdot (1 - R_1 R_2)^{-1} \{1 - R_2 R_3 \exp(-2\alpha) - 2 \sqrt{R_2 R_3} \exp[\cos \phi - \alpha]\}^{-1} \quad (1)$$

where α is optical absorption coefficient,

$$\begin{aligned} R_1 &= [(n_s - 1)/(n_s + 1)]^2, \\ R_2 &= [(n_f - n_s)/(n_f + n_s)]^2, \\ R_3 &= [(n_f - 1)/(n_f + 1)]^2, \\ \phi &= 4\pi n_f t / \lambda, \end{aligned}$$

n_s and n_f are the refractive index of substrate and films, respectively, t and λ are film thickness and wavelength. n_f can be obtained by

$$n_f = \frac{\lambda \lambda_i}{2t(\lambda - \lambda_i)} \quad (2)$$

where λ and λ_i are the wavelength corresponding to the i -th and $(i+1)$ extreme of the $T - \lambda$ curve. From Eqns. (1) and (2), absorption coefficient α is determined. The change of α with photon energy ($h\nu$) obeys the relation by

$$(h\nu)^2 = B(h\nu - E_{\text{opt}}) \quad (3)$$

where B is a constant and E_{opt} the optical gap energy. The E_{opt} values are gained by extrapolating the linear portion of the plots of $(2h\nu)^2$ vs $h\nu$ to α . Fig. 4 shows optical gap energy as a function of the annealing temperature for ZnO films (1.5% Al). The optical gap energy decreases from 3.75 eV to 3.68 eV with annealing temperature increasing from 25 °C to 400 °C.

The thermoelectric power of ZnO films was measured in the temperature range from 25 °C to 200 °C. Two copper-constantan thermocouples

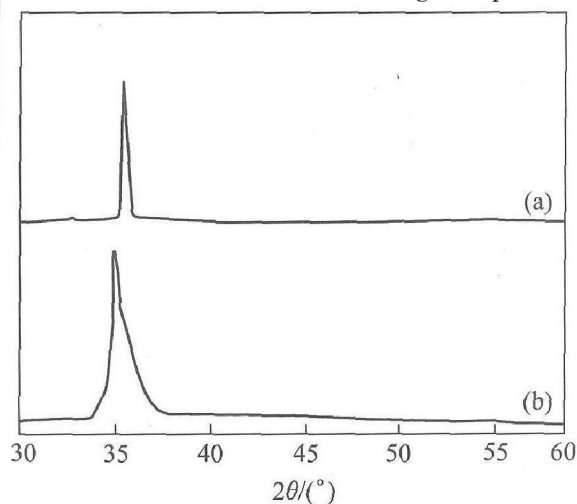


Fig. 2 Effect of annealing treatment on XRD patterns graphs of ZnO films
(a) —Annealing; (b) —Without annealing

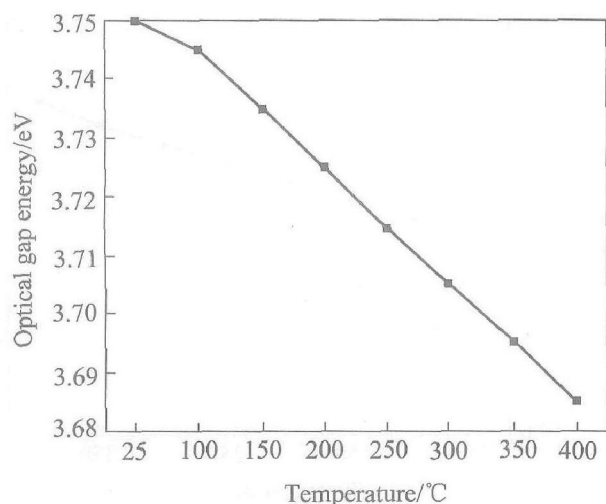


Fig. 4 Optical gap energy as function of annealing temperature for ZnO films

closely attached to the ends of a sample were used as measurement of both temperature difference ΔT and the thermal electromotive force in the films-copper circuit.

The reference points of the thermocouples were electrically insulated and put in an ice water bath. Typical plots of the thermoelectric power for ZnO films are shown in Fig. 5. It is very apparent that the annealing treatment could enhance Seebeck effect in the ZnO films.

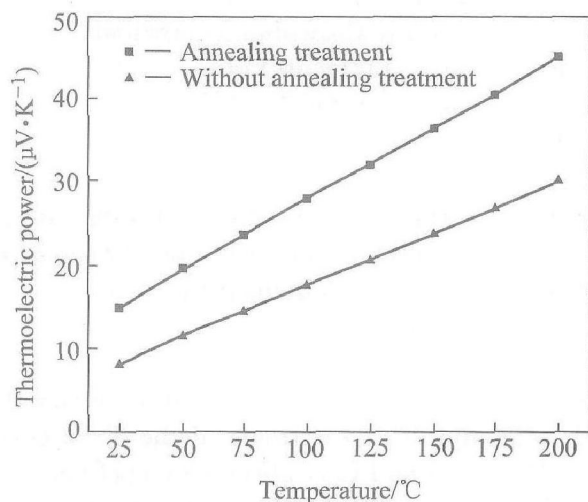


Fig. 5 Relationship between thermoelectric power and temperature for ZnO films

Like the experimental results mentioned above, the annealing treatment has an important influence on the optical and electrical properties of ZnO films. If one assumes spherical energy surface for the crystalline ZnO films, and that only the conduction band high curvature, then^[17, 18]

$$n = 8\pi/3h^3(m_e^* \Delta E)^{3/2} \quad (4)$$

where n is the free carrier concentration on the conduction band, ΔE is the Burstein shift (namely $\Delta E = E_{\text{opt}} - E_g$)^[17], and m_e^* is the effective mass of

conduction electron. The carrier concentration decreases and the resistivity increases after annealing treatment^[19], which causes ΔE to increase. Also the optical gap energy E_{opt} further decreases after heat treatment.

4 CONCLUSIONS

Al-doped ZnO films were prepared by R. F. magnetron sputtering. The experimental results show that the annealing treatment has a significant effect on the structure and properties of ZnO thin films. This is due to the decrease of carrier concentration after annealing treatment, resulting in Burstein shift. These results are very important for studying the quality and applications of ZnO films.

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